Sidebar 1

Under gradient conditions, the retention factor k varies, and needs to be written in a differential form:

$$k = \frac{dt_s}{dt_m}$$
 (S1.1)

 t_s is the time spent in the stationary phase, and t_m the time spent in the mobile phase. The elution pattern of a peak is obtained by integrating this equation over time:

$$\int_{0}^{t_{r}-t_{0}} \frac{1}{k} \cdot dt_{s} = \int_{0}^{t_{0}} dt_{m}$$
 (S1.2)

 t_r is the retention time. The retention factor k changes with the solvent composition c, which in turn changes linearly with time t:

$$k = k_0 \cdot e^{-S \cdot \Delta c \cdot t/t} g \qquad (S1.3)$$

 Δc is the difference in the solvent composition over the gradient run time t_g . Inserting this relationship into equation S1.2, we obtain:

$$\frac{1}{k_0} \cdot \int_0^{t_r - t_0} e^{S \cdot \Delta c \cdot t/t} dt = t_0 \quad (S1.4)$$

 t_0 is the retention time for an unretained peak, also called the column dead time. The integral can be solved to yield:

$$\frac{1}{k_0} \cdot \frac{t_g}{S \cdot \Delta c} \cdot \left(e^{S \cdot \Delta c \cdot \left(t_r - t_0 \right) / t_g} - 1 \right) = t_0 \quad (S1.5)$$

Rearranging this equation results in the expression for the retention factor under gradient conditions k_g :

$$k_g = \frac{t_r - t_0}{t_0} = \frac{1}{S \cdot \Delta c} \cdot \frac{t_g}{t_0} \cdot \ln\left(k_0 \cdot S \cdot \Delta c \cdot \frac{t_0}{t_g} + 1\right)$$
(S1.6)